



A Novel General Method for 2-Aminoglycal Synthesis

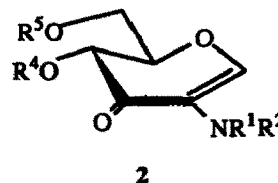
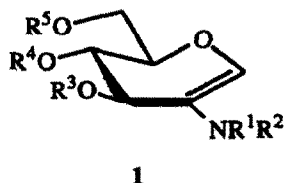
Fernando Iglesias-Guerra*, José I. Candela, José L. Espartero
and José M. Vega-Pérez*

Departamento de Química Orgánica y Farmacéutica, Facultad de Farmacia, Universidad de Sevilla.
41071 SEVILLA, Spain

Abstract: The first general method for *N,N*-substituted 2-aminoglycals is reported. We describe the syntheses of 2-[(*N*-acyl-*N*-alkyl)amino] and 2-[(*N,N*-dialkyl)amino]-1,5-anhydro-4,6-*O*-benzylidene-2-deoxy-*erythro*-hex-1-en-3-uloses by oxidation of glycoside derivatives of *N,N*-substituted-2-aminosugars using two oxidant systems. Reactions proceed with good yields.

Glycal derivatives have played a major role in carbohydrate chemistry as precursor substances in the synthesis of glycosides¹, thioglycosides², oligosaccharides³, 2-deoxynucleosides⁴ and C-glycosides⁵, as well as in that of α , β -unsaturated lactones⁶.

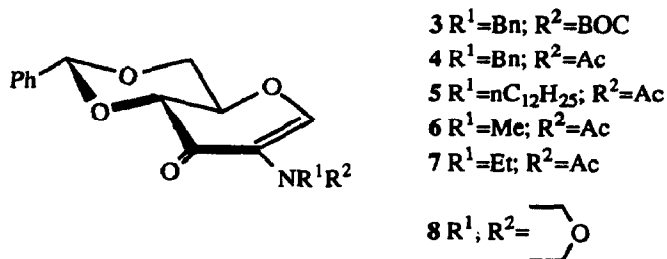
There are many works on glycal and 2-hydroxyglycal⁷ preparation, but many fewer on the synthesis of hex-1-en-3-ulose derivatives⁸. Moreover there are scarce antecedents for the preparation of 2-aminoglycal derivatives (1) and (2).



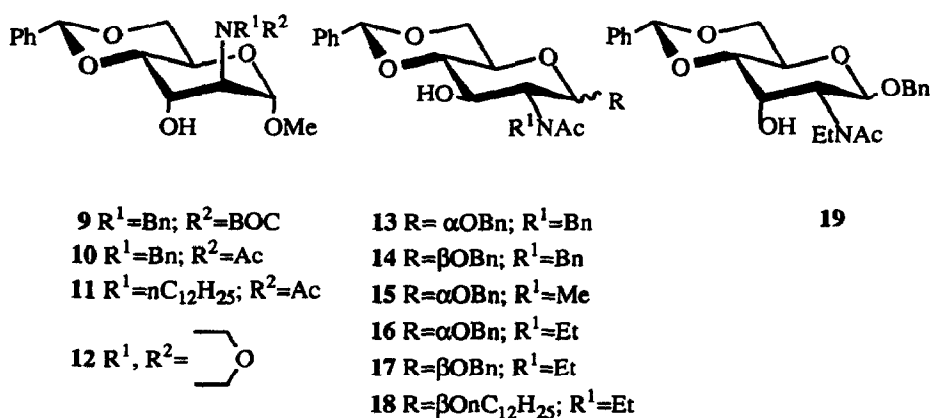
Such antecedents for the preparation of compounds 1 and 2 refer to non-general procedures of certain substances with acetyl or methyl group on the nitrogen⁹, or of other substances obtained as unwanted byproducts in glycosidation¹⁰ or α -methylation of ketones¹¹.

In view of the scarcity of the antecedents found and due to our interest in 2-aminoglycal derivatives, in this paper we describe for the first time a good general method for the synthesis of 2-[(*N*-acyl-*N*-alkyl)amino]-1,5-anhydro-4,6-*O*-benzylidene-2-deoxy-*erythro*-hex-1-en-3-uloses.

This general method consists of the oxidation of glycoside derivatives of *N,N*-substituted-2-aminosugars. Here, we describe the synthesis of substances 3-8 from 2-aminosugars (9-19)¹² using two



oxidant systems: PCC/CH₂Cl₂ and DMSO/DCC. A typical procedure for each reaction is given below.



The results of the oxidation reaction for the different compounds studied are recorded in Table 1. Analysis of the results reveals the general character of the procedure. It is noteworthy that the reactions give good yields using the two procedures, and byproducts are not detected. In addition, it is important to point out the ready preparation of 2-aminoglycal derivatives independently of the nature of the aglycon and nitrogen substituents and the configuration on the carbons C1, C2 and C3 of the starting aminosugar.

Table 1. 2-Aminoglycal derivatives by oxidation of *N,N*-substituted-2-aminoglycosides.

Entry	Starting material	Procedure	Product	Yield (%) ^a	MS(CI)	
					[M+H] ⁺	(% intensity)
1	9	A	3	80	438	(<5)
2	10	A	4	84	380	(100)
3	10	B	4	90	380	(100)
4	11	A	5	87	458	(100)
5	11	B	5	78	458	(100)
6	12	B	8	72	318	(100)
7	13 or 14	A	4	80-84	380	(100)
8	13 or 14	B	4	88-91	380	(100)
9	15	B	6	86	304	(100)
10	16 or 17	B	7	75-77	318	(100)
11	18	B	7	80	318	(100)
12	19	B	7	76	318	(100)

^aThe yield refers to the isolated yield.

These results contrast with those obtained in the oxidation of glycoside derivatives from glucose and altrose. Thus, when we oxidized (using the same procedures) methyl 4,6-*O*-benzylidene-2-*O*-pivaloyl- α -D-gluco- (or altro) pyranoside, the 3-ulose derivatives were obtained, with good yield, without 2-*O*-pivaloylglycal formation.

The products obtained were characterized by their elemental analyses, MS (EI and CI) and NMR data¹³.

Currently we are studying the reactivity of the obtained products with different nucleophiles.

Procedure A:

To a solution of 2-(*N*-acyl-*N*-alkyl)-aminosugar (2.0 mmol) in dry dichloromethane (40 mL) was added pyridinium chlorochromate (4 mmol) and 3 Å molecular sieves (10 g). The reaction mixture was well stirred overnight at room temperature. After completion of reaction (tlc), diethylether was added and filtered through a glass filter filled with silica gel containing CaSO₄ (10 %). Removal of the solvent by evaporation *in vacuo* gave a pure compound.

Procedure B:

To a stirred mixture of 2-(*N*-acyl-*N*-alkyl) or 2-(*N,N*-dialkyl)-aminosugar (2.0 mmol), 3 Å molecular sieves (1 g) and anhydrous dimethyl sulphoxide (30-40 mL), cooled to 0°, dicyclohexylcarbodiimide (8.0 mmol) was added. After ten minutes, anhydrous orthophosphoric acid (8.4 mmol) was then added portion-wise with cooling (ice bath) so that the temperature was kept at 25-30°. The reaction mixture was stirred for 24 hours at room temperature. The solids were removed by filtration and washed with dimethyl sulphoxide and acetone. The solution was diluted with four volumes of dichloromethane, water was added, and then a potassium carbonate solution (2.4 M) was added to bring the aqueous phase to about pH 8. The aqueous layer was extracted with dichloromethane and the combined extracts were washed with water until neutral. The organic phase was dried (Na₂SO₄) and evaporated *in vacuo* to dryness. The solid obtained was crystallized or fractionated by chromatography on a silica gel column.

Acknowledgements

We are grateful to the DGICYT of the Ministerio de Educación y Ciencia (Spain) for financial support (PB91-0620).

References and Notes

1. (a) Thiem, J.; Elvers, J. *Chem. Ber.* **1978**, *111*, 3514-3515. (b) Pelyvás, I.; Sztaricshai, F.; Bognar, R. *Carbohydr. Res.* **1979**, *76*, 257-260.
2. (a) Pelyvás, I.; Whistler, R.L. *Carbohydr. Res.* **1980**, *84*, C5-C7. (b) Pelyvás, I.; Hasegawa, A.; Whistler, R.L. *Carbohydr. Res.* **1986**, *146*, 193-203.
3. (a) Halcomb, R.; Danishefsky, S.J. *J. Am. Chem. Soc.* **1989**, *111*, 6661-6666. (b) Griffith, D.A.; Danishefsky, S.J. *J. Am. Chem. Soc.* **1990**, *112*, 5811-5819. (c) Griffith, D.A.; Danishefsky, S.J. *J. Am. Chem. Soc.* **1991**, *113*, 5863-5864.
4. Bowles, W.A.; Robins, R.K. *J. Am. Chem. Soc.* **1964**, *86*, 1252-1253.
5. Paulsen, H.; Bütsch, H. *Chem. Ber.* **1978**, *111*, 3484-3496.
6. (a) Rollin, P.; Sinay, P. *Carbohydr. Res.* **1981**, *98*, 139-142. (b) Jarglis, P.; Lichtenthaler, F.W. *Tetrahedron Lett.* **1982**, *23*, 3781-3784. (c) Lichtenthaler, F.W.; Ronninger, S.; Jarglis, P. *Liebigs Ann. Chem.* **1989**, 1153-1161.

7. Ferrier, R.J. In *The Carbohydrates*; Pigman and Horton Eds.; Academic Press: New York, 1980; Vol. 1B, pp. 843-879; and references cited therein.
8. (a) Collins, P.M. *Carbohydr. Res.* 1969, 11, 125-128. (b) Tronchet, J.M.J.; Tronchet, J.; Birkhäuser, A. *Helv. Chim. Acta* 1970, 53, 1489-1490. (c) Czernecki, S.; Vijayakumaran, K.; Ville, G. *J. Org. Chem.* 1986, 51, 5472-5475.
9. Pravdic, N.; Fletcher, H.G. Jr. *J. Org. Chem.* 1967, 32, 1806-1810. (b) Pravdic, N.; Franjic-Mihalic, I.; Danilov, B. *Carbohydr. Res.* 1975, 45, 302-306. (c) Baer, H.H.; Madumelu, C.B. *Carbohydr. Res.* 1978, 67, 329-340.
10. (a) Lemieux, R.U.; Takeda, B.; Chung, B.Y. *ACS Symp. Ser.* 1976, 39, 90-115. (b) Ogawa, T.; Nakabayashi, S.; Sasajima, K. *Carbohydr. Res.* 1981, 95, 308-312. (c) Gomtsyan, A.R.; Byramova, N.E.; Backinowsky, L.W.; Kochetkov, N.K. *Carbohydr. Res.* 1985, 138, C1-C4. (d) Maranduba, A.; Veyrières, A. *Carbohydr. Res.* 1986, 151, 105-119. (e) Marra, A.; Sinay, P. *Carbohydr. Res.* 1990, 200, 319-337. (f) Maloisel, J.L.; Vasella, A. *Helv. Chim. Acta* 1992, 75, 1491-1514.
11. Klemer, A.; Wilbers, H. *Liebig Ann. Chem.* 1985, 2328-2341.
12. For the synthesis of new starting material (9-19), described methods are used¹⁴; such products will be described in a forthcoming paper.
13. For 4: (Found: C, 69.43; H, 5.76; N, 3.74; C₂₂H₂₁NO₅ requires C, 69.65; H, 5.58; N, 3.69). δ_{H} (500 MHz, DMSO-d₆, 25°) 7.62(s), 7.59(s) (1H, H-1), 4.94(d), 4.77(d) (1H, H-4), 4.68(m) (1H, H-5), 4.41(m) (1H, H-6_{eq}), 4.06(m) (1H, H-6_{ax}), 5.71(s), 5.69(s) (1H, PhCH), 5.08(dd), 3.96(dd) (2H, PhCH₂N), 1.98(s), 1.85(s) (3H, CH₃). δ_{C} (50Mz, DMSO-d₆, 25°) 162.3, 161.9 (C-1), 120.5, 120.1 (C-2), 185.1, 184.8 (C-3), 75.1 (C-4), 71.9 (C-5), 65.9 (C-6), 100.0, 99.8 (PhCH), 50.4, 48.6 (PhCH₂N), 170.4, 169.9 (CON), 20.9, 20.7 (CH₃). The assignment was confirmed by DEPT, COSY and CHCORR experiments. When we recorded the spectrum at 110° all double signals collapsed to single signals.
14. (a) Chini, M.; Crotti, P.; Macchia, F. *Tetrahedron Lett.* 1990, 31, 4661-4664. (b) Vega-Pérez, J.M.; Espartero, J.L.; Alcudia, F. *Carbohydr. Res.* 1992, 235, C5-C7. (c) Vega-Pérez, J.M.; Espartero, J.L.; Alcudia, F. *J. Carbohydr. Chem.* 1993, 12, 477-486.

(Received in UK 10 March 1994; accepted 12 May 1994)